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reaction can be concisely represented by a chemical 1

The substances that undergo a chemical change are the

2 The new substances formed in a chemical reaction

are the 3

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chemical equation. skeleton equation. catalyst.

coefficients. a representation of the chemical reaction;

the formulas of the.... a chemical equation that does not

indicate the relative amount.... a substance that speeds

up a reaction but is not used up in th....

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a chemical change in which one element replaces a
second element in a compound; displacement reaction
Activity Series a list of elements in order of decreasing
activity

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Stoker, Introduction to Chemical Principles | Pearson

- A skeleton equation is a chemical equation that does not indicate the relative amounts of the reactants and products. Like a recipe with no amounts listed.

Chapter 11

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11.2 Types of Chemical Reactions > 13 A

decomposition reaction is a chemical change in which a single compound breaks down into two or more simpler products.

- Decomposition reactions involve only one reactant and two or more products.
- The products can be any combination of elements and compounds.
- Most decomposition reactions require energy

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