

## Mole Lab Answers

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Mole Lab continued 2 2017 linn scientiic Inc ll iht eered Write the actual mass here 0.98 x (correct answer) 0.95 x (correct answer) 0.90 x (correct answer) 1.02 x (correct answer) 1.05 x (correct answer) 1.10 x (correct answer) Correct answer: (actual mass divided by 63.546 times 6.022e23) 4 3 2 Figure 1. Procedure 1.

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The fastest way to obtain a relative mass of beans would be to count the beans. The fastest way to obtain a mole of beans would be to weigh them. (At least in principle. The mass of a mole of beans would be incredibly large- on the order of  $10^{22}$  g.) Part III. All atomic masses agree with the relative masses to three significant figures.

## *Laboratory Activity 1: Teacher Notes Continued*

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Have a conceptual understanding of the mole as the method of "counting" items and finding the mass of items that can't be seen. 2. Be able to calculate the number of items (molecules, atoms, ions, and formula units) if given the number of moles. 3. Be able to calculate the number of moles if given the number of items.

## *#20 Introduction to the Mole - Terrific Science*

Answer # target gas law lab continue 5 216 linn scientiic inc all ights esered target gas law lab worksheet many chemical reactions involve gaseous reactants or products the volumes of gases that react or are produced depend entirely on the moles of the gas and of course on the pressure and

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